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By Shiksha House ~~Chemistry Chapter 11 Standardized Test~~

Use the following passage and table to answer questions

11–14. A student studying electron configurations in atoms

created Table 1 in order to summarize the various facts she

learned about principal energy levels, energy sublevels, and

orbitals. 11. Based on Table 1 what is the relationship

between the principal energy level number (n)

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substances varied with temperature she tested.

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The study of quantitative relationships between the amounts of.... In a balanced equation, the ratio between the numbers of moles.... A reactant that remains after a chemical reaction stops. A reactant that is totally consumed during a chemical reaction....

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