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Chemistry a molecular approach: by Nivaldo J Tro Print book: English, 2020. Fifth edition : Hoboken, NJ Pearson Education, Inc 2. Chemistry : a molecular approach: 2. Chemistry : a molecular approach. by Nivaldo J Tro; John Bertram Vincent; Erica Livingston Print book:

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Dr. Tro has received Westmont s outstanding teacher of the year award three times. He is the successful author of the market leader text, "Introductory Chemistry," 5e (Pearson), "Chemistry: A Molecular Approach," 3e (Pearson), "Chemistry: Structure and Properties," 1e (Pearson), and "Chemistry In Focus: A Molecular View of Our World," 4e.

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Chemistry: A Molecular Approach, Second Canadian Edition Plus Mastering Chemistry with Pearson eText -- Access Card Package (2nd Edition) Hardcover | Feb. 25 2016 by Nivaldo J. Tro (Author), Travis D. Fridgen (Author), Lawton E. Shaw (Author) 4.7 out of 5 stars 15 ratings See all formats and editions

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Description With a focus on real-world applications and a conversational tone, this laboratory manual contains 28 experiments written specifically to correspond with Chemistry: A Molecular Approach, Second Edition by Nivaldo J. Tro. Each experiment covers one or more topics discussed within a chapter of the textbook, with the dual goal of 1) helping students understand the underlying concepts ...

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This innovative, pedagogically driven text explains difficult concepts in a student-oriented manner. The book offers a rigorous and accessible treatment of general chemistry in the context of relevance. Chemistry is presented visually through multi-level images--macroscopic, molecular and symbolic representations--helping students see the connections among the formulas (symbolic), the world around them (macroscopic), and the atoms and molecules that make up the world (molecular). KEY TOPICS: Units of Measurement for Physical and Chemical Change;Atoms and Elements;Molecules, Compounds, and Nomenclature;Chemical Reactions and Stoichiometry;Gases;Thermochemistry;The Quantum-Mechanical Model of the Atom;Periodic Properties of the Elements;Chemical Bonding I: Lewis Theory;Chemical Bonding II: Molecular Shapes, Valence Bond Theory, and Molecular Orbital Theory;Liquids, Solids, and Intermolecular Forces;Solutions;Chemical Kinetics;Chemical Equilibrium;Acids and Bases;Aqueous Ionic Equilibrium;Gibbs Energy and Thermodynamics;Electrochemistry;Radioactivity and Nuclear Chemistry;Organic Chemistry I: Structures;Organic Chemistry II: Reactions;Biochemistry;Chemistry of the Nonmetals;Metals and Metallurgy;Transition Metals and Coordination Compounds MARKET: Appropriate for General Chemistry (2 - Semester) courses.

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Adapted from Nivaldo J. Tro's best-selling general chemistry book, Principles of Chemistry: A Molecular Approach focuses exclusively on the core concepts of general chemistry without sacrificing depth or relevance. Tro's unprecedented two- and three-column problem-solving approach is used throughout to give students sufficient practice in this fundamental skill. A unique integration of macroscopic, molecular, and symbolic illustrations helps students to visualize the various dimensions of chemistry; Tro's engaging writing style captures student's attention with relevant applications. The Second Edition offers a wealth of new and revised problems, approximately 50 new conceptual connections, an updated art program throughout, and is available with MasteringChemistry®, the most advanced online tutorial and assessment program available. This package contains: Principles of Chemistry: A Molecular Approach, Second Edition

Following up on her acclaimed Teach Students How to Learn, that describes teaching strategies to facilitate dramatic improvements in student learning and success, Sandra McGuire here presents these [secrets] direct to students. Her message is that [Any student can use simple, straightforward strategies to start making A's in their courses and enjoy a lifetime of deep, effective learning.] Beginning with explaining how expectations about learning, and the study efforts required, differ between college and secondary school, the author introduces her readers, through the concept of metacognition, to the importance and powerful consequences of understanding themselves as learners. This framework and the recommended strategies that support it are useful for anyone moving on to a more advanced stage of education, so this book also has an intended audience of students preparing to go to high school, graduate school, or professional school. In a conversational tone, and liberally illustrated by anecdotes of past students, the author combines introducing readers to concepts like Bloom's Taxonomy (to illuminate the difference between studying and learning), fixed and growth mindsets, as well as to what brain science has to tell us about rest, nutrition and exercise, together with such highly specific learning strategies as how to read a textbook, manage their time and take tests. With engaging exercises and thought-provoking reflections, this book is an ideal motivational and practical text for study skills and first year experience courses.

The authors, who have more than two decades of combined experience teaching an atoms-first course, have gone beyond reorganizing the topics. They emphasize the particulate nature of matter throughout the book in the text, art, and problems, while placing the chemistry in a biological, environmental, or geological context. The authors use a consistent problem-solving model and provide students with ample opportunities to practice.

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