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For students using the Foundation edition, assign problems 1:112, 15:116. It is the mass of 1 mol of that compound. 80.1 48.0 The mass of a mole of a compound is determined by adding the atomic masses of the atoms making up the molecule. The atomic mass of an element is the mass of a single atom in atomic mass units.

Chemical Quantities - AP Biology
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Chapter 10 Chemical Quantities Quiz Answer
CHAPTER 10: Chemical Quantities BASICS: ∴ The basic unit that is used to determine the amount of a chemical substance is called a mole ∴ A mole(mol) of a substance is equivalent to 6.02 x 1023 particles of that